

4'-Fluorochalcone

Shea-Lin Ng,^a Ibrahim Abdul Razak,^a Hoong-Kun Fun,^{a*} P. S. Patil^b and S. M. Dharmaprakash^b

^aX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia, and ^bDepartment of Studies in Physics, Mangalore University, Mangalagangothri, Mangalore 574 199, India

Correspondence e-mail: hkfun@usm.my

Key indicators

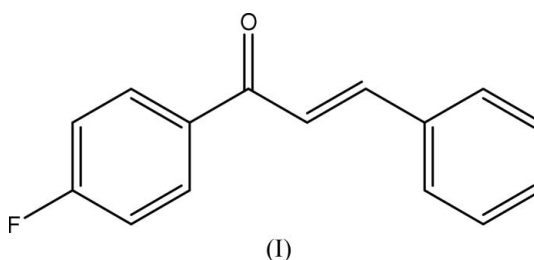
Single-crystal X-ray study
 $T = 100$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
 R factor = 0.049
 wR factor = 0.154
Data-to-parameter ratio = 16.5

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The enone fragment and the benzene rings of the title compound, $\text{C}_{15}\text{H}_{11}\text{FO}$, are each planar. The crystal packing is stabilized by weak intermolecular $\text{C}-\text{H}\cdots\pi$ interactions involving both aromatic rings; the molecules are stacked along the b axis.

Comment

The interest in chalcone derivatives in several disciplines stems from their biological activities, including antifungal (Boeck *et al.*, 2005) and anticoagulant properties (Shuib *et al.*, 1999), and their pharmacological activities, such as anti-protozoal (Nielsen *et al.*, 1998; Li *et al.*, 1995; Liu *et al.*, 2001), anti-inflammatory (Hsieh *et al.*, 1998), nitric oxide inhibition (Rojas *et al.*, 2002) and anticancer properties. The compounds are also used as depigmenting agents (Khatib *et al.*, 2005). Recently, it has been noted that derivatives of chalcones exhibit extremely high and fast non-linearity (Fichou *et al.*, 1988; Kitaoka *et al.*, 1990; Uchida *et al.*, 1998; Goto *et al.*, 1991; Patil *et al.*, 2006*a,b*; Zhang *et al.*, 1990; Zhao *et al.*, 2000). In view of these features associated with chalcones, we and others have undertaken a number of theoretical and structural studies of such compounds (Ng, Patil *et al.*, 2006; Ng, Shettigar *et al.*, 2006; Patil *et al.*, 2006*a,b*; Teh *et al.*, 2006; Radha Krishna *et al.*, 2005; Sathiya Moorthi *et al.*, 2005; Uchida *et al.*, 1995), and we report here the structure of the title compound, (I) (Fig. 1). Crystals of (I) do not exhibit second-order non-linear optical properties as they crystallize in a centrosymmetric space group.



The bond lengths and angles in (I) display normal values (Allen *et al.*, 1987) and agree well with those observed in related structures (Ng, Patil *et al.*, 2006; Ng, Shettigar *et al.*, 2006; Patil *et al.*, 2006*a,b*).

The enone group (O1/C7–C9) and the two benzene rings (C1–C6 and C10–C15) of the chalcone are each planar, with maximum deviations of 0.047 (2), 0.010 (2) and 0.006 (2) Å for atoms C7, C2 and C13, respectively.

The molecule is twisted about the C6–C7 bond, with a dihedral angle of 46.75 (5)° between the two benzene rings.

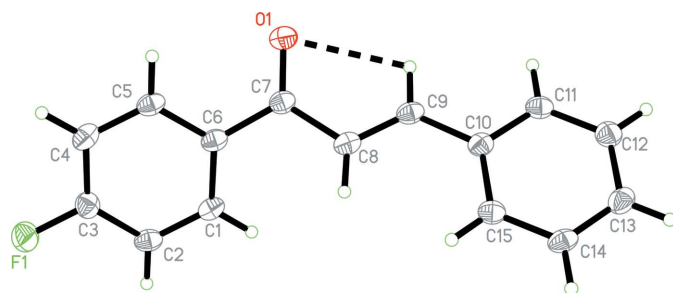


Figure 1
View of (I), showing the atomic numbering scheme and 50% probability displacement ellipsoids. The dashed line indicates a hydrogen bond.

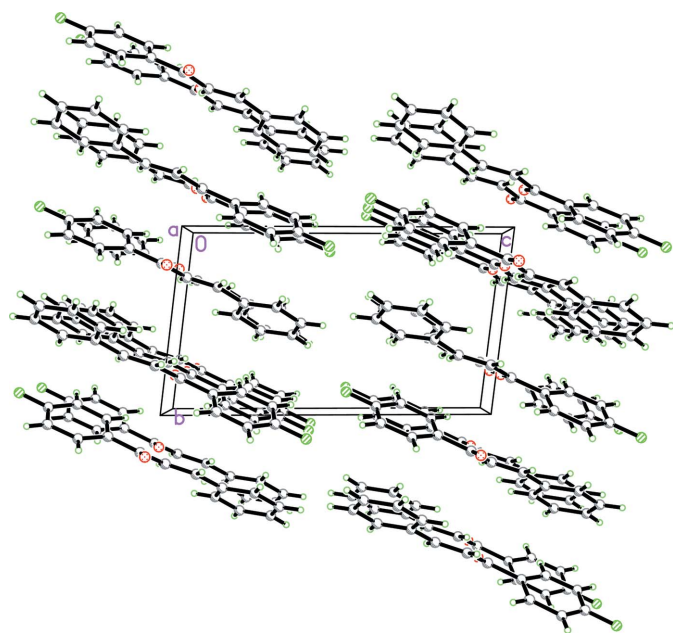


Figure 2
The crystal packing of (I), viewed down the *a* axis.

The least-squares plane through the enone fragment makes dihedral angles of 27.16 (7)° with the C1–C6 benzene ring and 20.24 (7)° with the C10–C15 benzene ring.

In the molecule, an intramolecular C9–H9A···O1 interaction generates an *S*(5) ring motif (Bernstein *et al.*, 1995). Aromatic rings C1–C6 (centroid *Cg*1) and C10–C15 (centroid *Cg*2) are involved in weak intermolecular C–H··· π interactions (Table 1) which stabilize the crystal structure. The molecules are stacked along the *b* axis (Fig. 2).

Experimental

Benzaldehyde (0.01 mol) and 4-fluoroacetophenone (0.01 mol) were stirred in ethanol (60 ml) at room temperature. An aqueous solution of NaOH (10 ml, 30%) was added and the mixture was stirred for 4 h. The precipitate which formed was filtered, washed with water and dried. The resulting crude product was recrystallized twice from acetone. Crystals of (I) suitable for a single-crystal diffraction study were grown by slow evaporation of an acetone solution.

Crystal data

$C_{15}H_{11}FO$
 $M_r = 226.24$
Triclinic, $P\bar{1}$
 $a = 5.8391$ (2) Å
 $b = 7.4435$ (3) Å
 $c = 13.0358$ (5) Å
 $\alpha = 96.592$ (2)°
 $\beta = 93.947$ (2)°
 $\gamma = 93.593$ (2)°

$V = 560.07$ (4) Å³
 $Z = 2$
 $D_x = 1.342$ Mg m⁻³
Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 100.0$ (1) K
Plate, yellow
0.38 × 0.30 × 0.10 mm

Data collection

Bruker SMART APEX2 CCD area-detector diffractometer
 ω scans
Absorption correction: multi-scan (*SADABS*; Bruker, 2005)
 $T_{\min} = 0.847$, $T_{\max} = 0.991$

6883 measured reflections
2538 independent reflections
1843 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.049$
 $\theta_{\max} = 27.5^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.049$
 $wR(F^2) = 0.154$
 $S = 1.05$
2538 reflections
154 parameters

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0877P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.27$ e Å⁻³
 $\Delta\rho_{\min} = -0.26$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
C2–H2A··· <i>Cg</i> 2 ⁱ	0.93	2.90	3.528 (2)	126
C5–H5A··· <i>Cg</i> 2 ⁱⁱ	0.93	2.84	3.442 (2)	123
C9–H9A··· <i>Cg</i> 1 ⁱⁱⁱ	0.93	3.08	3.639 (2)	120
C14–H14A··· <i>Cg</i> 1 ^{iv}	0.93	2.87	3.536 (2)	129
C9–H9A···O1	0.93	2.48	2.814 (2)	101

Symmetry codes: (i) $-x + 1, -y, -z$; (ii) $-x + 2, -y + 1, -z$; (iii) $-x + 2, -y, -z$; (iv) $-x + 1, -y + 1, -z$.

H atoms were placed in calculated positions and constrained to ride on their carrier atoms, with C–H = 0.93 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: *APEX2* (Bruker, 2005); cell refinement: *APEX2*; data reduction: *SAINT* (Bruker, 2005); program(s) used to solve structure: *SHELXTL* (Sheldrick, 1998); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 2003).

The authors thank the Malaysian Government and Universiti Sains Malaysia for the Scientific Advancement Grant Allocation (SAGA) grant No. 304/PFIZIK/653003/A118 and the USM short-term grant No. 304/PFIZIK/635028. PSP and SMD are grateful to DRDO, Government of India, for financial assistance.

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.

- Boeck, P., Leal, P. C., Yunes, R. A., Filho, V. C., Lopez, S., Sortino, M., Escalante, A., Furlan, R. L. E. & Zacchino, S. (2005). *Arch. Pharm. Chem. Life Sci.* **338**, 87–95.
- Bruker (2005). *APEX2* (Version 1.27), *SAINT* (Version V7.12A) and *SADABS* (Version 2004/1). Bruker AXS Inc., Madison, Wisconsin, USA.
- Fichou, D., Watanabe, T., Takeda, T., Miyata, S., Goto, Y. & Nakayama, M. (1988). *Jpn J. Appl. Phys.* **27**, L429–L430.
- Goto, Y., Hayashi, A., Kimura, Y. & Nakayama, M. (1991). *J. Cryst. Growth*, **108**, 688–698.
- Hsieh, H. K., Lee, T. H., Wang, J. P., Wang, J. J. & Lin, C. N. (1998). *Pharm. Res.* **15**, 39–46.
- Khatib, S., Nerya, O., Musa, R., Shmuel, M., Tamir, S. & Vaya, J. (2005). *Bioorg. Med. Chem.* **13**, 433–441.
- Kitaoka, Y., Sasaki, T., Nakai, S., Yokotani, A., Goto, Y. & Nakayama, M. (1990). *Appl. Phys. Lett.* **56**, 2074–2076.
- Li, R., Kenyon, G. L., Cohen, F. E., Chem, X., Gong, B., Dominguez, J. N., Davidson, E., Kurzban, G., Miller, R. E., Nuzum, E. O., Rosenthal, P. J. & McKerrow, J. H. (1995). *J. Med. Chem.* **38**, 5031–5037.
- Liu, M., Wilairat, P. & Go, M. L. (2001). *J. Med. Chem.* **44**, 4443–4452.
- Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.
- Ng, S. L., Patil, P. S., Razak, I. A., Fun, H. K. & Dharmaparakash, S. M. (2006). *Acta Cryst.* **E62**, o893–o895.
- Ng, S. L., Shettigar, V., Razak, I. A., Fun, H. K., Patil, P. S. & Dharmaparakash, S. M. (2006). *Acta Cryst.* **E62**, o1421–1423.
- Nielsen, S. F., Christensen, S. B., Cruciani, G., Kharazmi, A. & Liljefors, T. (1998). *J. Med. Chem.* **41**, 4819–4832.
- Patil, P. S., Teh, J. B.-J., Fun, H.-K., Razak, I. A. & Dharmaparakash, S. M. (2006a). *Acta Cryst.* **E62**, o896–o898.
- Patil, P. S., Teh, J. B.-J., Fun, H.-K., Razak, I. A. & Dharmaparakash, S. M. (2006b). *Acta Cryst.* **E62**, o1710–o1712.
- Radha Krishna, J., Jagadeesh Kumar, N., Krishnaiah, M., Venkata Rao, C., Koteswara Rao, Y. & Puranik, V. G. (2005). *Acta Cryst.* **E61**, o1323–o1325.
- Rojas, J., Payá, M., Domínguez, J. N. & Ferrándiz, M. L. (2002). *Bioorg. Med. Chem. Lett.* **12**, 1951–1954.
- Sathya Moorthi, S., Chinnakali, K., Nanjundan, S., Radhika, R., Fun, H.-K. & Yu, X.-L. (2005). *Acta Cryst.* **E61**, o480–o482.
- Sheldrick, G. M. (1998). *SHELXTL*. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Shuib, N. S., Sam, T. W., Wong, K. C., Chinnakali, K. & Fun, H.-K. (1999). *Acta Cryst.* **C55**, 576–578.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Teh, J. B.-J., Patil, P. S., Fun, H.-K., Razak, I. A. & Dharmaparakash, S. M. (2006). *Acta Cryst.* **E62**, o2261–o2262.
- Uchida, T., Kozawa, K., Kimura, Y. & Goto, Y. (1995). *Synth. Met.* **71**, 1705–1706.
- Uchida, T., Kozawa, K., Sakai, T., Aoki, M., Yoguchi, H., Abdureyim, A. & Watanabe, Y. (1998). *Mol. Cryst. Liq. Cryst.* **314**, 135–140.
- Zhang, G., Kinoshita, T., Sasaki, K., Goto, Y. & Nakayama, M. (1990). *J. Cryst. Growth*, **100**, 411–416.
- Zhao, B., Lu, W.-Q., Zhou, Z.-H. & Wu, Y. (2000). *J. Mater. Chem.* **10**, 1513–1517.